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New Macrocyclic Ligands with P₂N₄S₂, Rings in the Framework

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NEW MACROCYCLIC LIGANDS WITH P₂N₄S₂ RINGS IN THE FRAMEWORK.

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<u>Abstract</u> Several new macromolecules containing two Ph₄P₂N₄S₂ rings linked by meta-xylyl groups have been prepared.

INTRODUCTION

Over the last three years we have developed simple routes to inorganic heterocycles of the type $[Ph_4P_2N_4(SR)(SR')]^1$ which employ the reactive alkali metal reagents $[Li(Ph_4P_2N_4S_2R)(THF)]_2^2$ and $M_2[Ph_4P_2N_4S_2]$ $(M=Li, Na).^3$

$$M_{2}[Ph_{4}P_{2}N_{4}S_{2}] + 2RI \xrightarrow{-2MI} Ph_{2}P \xrightarrow{N} PPh_{2}$$

$$(M = Li, Na)$$

$$(R = Me. allvl)$$

$$\frac{1}{2}[\text{Li}(\text{Ph}_{4}\text{P}_{2}\text{N}_{4}\text{S}_{2}\text{R}')]_{2} + \text{RI} \xrightarrow{-\text{LiI}} \text{Ph}_{2}\text{P}' \underbrace{\begin{array}{c} R' \\ S \\ N \\ S \\ R \end{array}} \text{PPh}_{2}$$

$$(2)$$

$$(R = \text{Me, allyl; } R' = {}^{\text{t}}\text{Bu, Me, Ph})$$

Very recently we began using these versatile metal reagents to construct novel macromolecules composed of $P_2N_4S_2$ rings and meta-xylyl building blocks. These supramolecular systems were prepared with the ultimate aim of employing them as host

molecules for metal cations from both sides of the periodic table.

SYNTHESIS AND CHARACTERIZATION OF 1, 2 AND 3

The neutral molecules 1a and 1b were first prepared by the reaction of $\underline{\text{meta}}$ - α , α' -dibromoxylene with $[\text{Li}(Ph_4P_2N_4S_2R)]_2$ (R = Me, tBu). The dipotassium salt of the dianionic compound 2 was synthesized by mixing $[\text{meta-}(KCH_2)_2C_6H_4]$ with two equiv. of 1,5- $[Ph_4P_2N_4S_2]$ in THF at -78°C. Compound 2 is readily converted to 1a by treatment with 2 equiv. of iodomethane. The macrocycle 3 is most conveniently prepared by the reaction of $K_2[Ph_4P_2N_4S_2]$ with an equimolar quantity of $\underline{\text{meta-}}\alpha$, α' -dibromoxylene. This compound is also accessible via the treatment of the dipotassium salt of 2 with one equiv. of $\underline{\text{meta-}}\alpha$, α' -dibromoxylene.

Compounds 1a, 1b and 3 were characterized by elemental analysis, FAB mass spectroscopy and ¹H/¹³C/³¹P NMR spectroscopy. The identity of 2 was established on the basis of ³¹P NMR spectroscopy and its derivative chemistry.

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